

# Rechargeable solid state battery

OverviewHistoryMaterialsUsesChallengesAdvantagesThin-film solid-state batteriesInnovation and IP protectionBetween 1831 and 1834, Michael Faraday discovered the solid electrolytes silver sulfide and lead(II) fluoride, which laid the foundation for solid-state ionics. By the late 1950s, several silver-conducting electrochemical systems employed solid electrolytes, at the price of low energy density and cell voltages, and high internal resistance. In 1967, the discovery of fast ionic conduction  $\gamma$ -alumina for a broad class of ions ( $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Ag}^+$ , and R...

Thus, regarding rechargeable lithium ion batteries, it is also possible to compose a battery with excellent safety by an all-solid-state battery using a solid electrolyte. Using solid electrolytes for lithium-ion batteries can improve their ...

Solid-state batteries (SSBs) offer significant improvements in safety, energy density, and cycle life over conventional lithium-ion batteries, with promising applications in electric vehicles and grid storage due to their non ...

A reversible plating/stripping of a dendrite-free metallic-sodium anode with a reduced anode/ceramic interfacial resistance is created by a thin interfacial interlayer formed in situ or by the introduction of a dry polymer film. ...

CeraCharge(TM), the world's first rechargeable all-ceramic solid-state SMD battery, offers high energy density, miniaturization, and a high degree of safety with no risk of electrolyte leakage. These outstanding features were realized thanks to ...

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