

# Block copolymer solid battery electrolyte with high Li-ion transference number

What are the characteristics of a block copolymer electrolyte?

The optimized block copolymer electrolyte has shown an amorphous crystalline structure, a high ionic conductivity of  $\sim 5.7 \times 10^{-4} \text{ S cm}^{-1}$ , high lithium ion transference number ( $\sim 0.49$ ), wide electrochemical window up to  $\sim 4.65 \text{ V}$  (vs.  $\text{Li}^+/\text{Li}$ ) and favorable mechanical strength at  $55^\circ\text{C}$ .

Are solid polymer electrolytes suitable for Li-based rechargeable batteries?

Recent advancements in solid polymer electrolytes (SPEs) for Li-based rechargeable batteries are summarized. The principles of Li-ion conduction inside SPE and the corresponding strategies to improve Li-ion conductivity and transference number are first reviewed.

Is block copolymer electrolyte a homogeneous conductor?

Different from the blend/composite SPE, the block copolymer electrolyte (BCPE) has been proposed as homogeneous conductor from different polymer chains for the  $\text{Li}^+$  migration. However, the comprehensive properties of BCPE are still not satisfactory for LMBS.

Can solid polymer electrolytes improve the safety of high-energy lithium metal batteries?

Solid polymer electrolytes have been considered as the promising candidate to improve the safety and stability of high-energy lithium metal batteries.

Which solid electrolyte should be used in LMBS?

The ideal solid electrolyte used in LMBS should not only have high ionic conductivity ( $\sigma$ ) and high ion transference number ( $t_{\text{Li}^+}$ ) within a certain temperature range [12], but also be a good  $\text{Li}^+$  conductor and electronic insulator [13].

How can single ion polymer electrolytes improve battery performance?

During this process, high Li concentration gradients form in the battery system, which often leads to uneven  $\text{Li}^+$  deposition on the Li metal surface. Implementation of single-ion polymer electrolytes in batteries can effectively reduce this concentration gradient and inhibit the Li dendrite formation.



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